

Assessment Scheme

SECTION - I

Q 1 Attempt all questions. Each question is of one mark. 20

1. Electrodialysis is used for _____ of colloids.
[A] Preparation
[B] Identification
[C] Purification
[D] Stabilization

2. Colloidal dispersion of Sulphur is _____ type of colloids.
[A] Lyophobic
[B] Hydrophili
[C] Lyophilic
[D] Association

3. Zeta potential value for stable dispersion is _____.
[A] ± 20 to ± 50
[B] +10 to -10
[C] +20 to -20
[D] +5 to -5

4. Charcoal powder on water surface has _____ contact angle (In degrees)
[A] 0
[B] 90
[C] 180
[D] 270

5. Flocalated suspension has _____
[A] Strong inter particulate attaction
[B] Weak interparticulate attraction
[C] Strong interparticulate repulsion
[D] Weak inter particulate trepulsion

6. Structured vehicle is used in suspension to _____.
[A] Decrease interfacial tension
[B] To prevent caking of sediment
[C] To prevent sedimentation of particles
[D] To reduce size

7. HLB value for emulsifying agent to prepare O/W emulsion is _____
[A] 1-5
[B] 2-4
[C] 9-16
[D] 1-8
8. _____ emulsifying agent shows solid particle adsorption.
[A] Accacia
[B] Tragacanth
[C] Bentonite
[D] Methyl cellulose
9. On commercial scale, emulsion is prepared by _____.
[A] Centrifugation
[B] dialysis
[C] Freezing
[D] Homogenization
10. O/W type of emulsion will give _____ creaming, W/O type of emulsion will give _____ creaming
[A] Upward, Downward
[B] Downward, Upward
[C] Downward, Middle
[D] Upward, Middle
11. EDTA is _____
[A] Divalent
[B] Pentavalent
[C] Tetravalent
[D] Hexavalent
12. Cisplatin has _____ platinum.
[A] Divalent
[B] Trivalent
[C] tetravalent
[D] Pentavalent
13. Negative thixotropy is as similar as _____.
[A] Plastic flow
[B] Newtonian flow
[C] Non Newtonian flow
[D] Rheopexy
14. Brookfield viscometer is an example of _____ type.
[A] Cone and plate
[B] Extrusion
[C] Rotating sphere
[D] Rotating spindle

15. Andersen apparatus consist of _____
[A] Balance
[B] Electrodes
[C] Pipette
[D] Hydrometer
16. If true density of talc is 3.00 g/cm^3 , then bulk density of talc is _____ g/cm^3 .
[A] Greater than 3.00 g/cm^3
[B] Equal to 3.00 g/cm^3
[C] less than 3.00 g/cm^3
[D] No correlation
17. Granule density of Aspirin is 1.33 g/cm^3 , true density of the same is 1.37 g/cm^3 . Calculate the interparticle porosity.
[A] 5.6 g/cm^3
[B] 10.2 g/cm^3
[C] 2.9 g/cm^3
[D] 4.5 g/cm^3
18. Human blood has _____ pH.
[A] 4.5
[B] 2.5
[C] 7.4
[D] 8.5
19. Colloids with smaller particle size shows _____ color.
[A] Red
[B] Yellow
[C] Blue
[D] Black
20. For non porous powder, granule density and true density will be _____
[A] Different
[B] Same
[C] True density is greater
[D] Granule density is greater

SECTION – II

- Q 2** Attempt any **TWO** of the following
- A** Enlist the methods to determine particle size of powder and explain any one in detail. **05**
- B** Discuss the two component liquid system showing upper consolute temperature. **05**
- C** Describe the methods of preparation for Lyophobic colloids. **05**
- Q 3** Attempt any **TWO** of the following
- A** Define true density and explain the methods to determine the true density in detail. **05**
- B** Discuss about the physical stability of suspension. **05**
- C** Describe the mechanism of action for emulsifying agents. **05**

SECTION – III

- Q 4** Attempt any **FOUR** of the following
- A** Enlist the methods for determination of surface and interfacial tension. Explain capillary rise method. **05**
- B** Write a note on Hydrophilic Lipophilic Balance. **05**
- C** Define rheology. Give the application of Rheology in Pharmacy. **05**
- D** Enumerate methods to determine viscosity. Explain Oswald method. **05**
- E** Write a note on protective colloids. **05**
- F** Classify types of complex. Explain inclusion complex in detail. **05**
- Q 5** Attempt any **FOUR** of the following
- A** Define surface tension. Explain Du-Nouy Ring method. **05**
- B**
- 1.** Calculate Specific surface (S_w & S_v) for griseofulvin having d_{vs} 3.0 micron and true density 1.455 g/cm^3 . **03**
- 2.** What are the factors affecting solubility of gases in liquids? **02**
- C** Explain Non Newtonian flow with rheogram, and with suitable example. **05**
- D** Write application of complexes in dosage forms. **05**
- E** Write note on metal complexes. **05**
- F** Describe in brief about preparations and stability of isotonic solutions. **05**
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